

The Impact of Social Motivational, Cognitive, and Pedagogic Content Dimensions  
on Understanding Chemistry Concepts

Dr. Safa A. Zaid-Elkilani

University of Jordan

Department of Curriculum & Instruction

Jordan

Dr. Husein F. Ramzoun

Al-Zaytouna University

College of Literacy

Jordan

Abstract

The study examines the impact of social motivational, cognitive, and pedagogic content dimensions on

understanding chemistry problems. Three groups of university students: freshman science college students, n

( ), freshman inservice students' teachers n ( ), Junior and Senior inservice students' teachers n( ), had been

examined in understanding chemistry concepts. The results show that the social motivation, cognitive, and

pedagogic dimensions have a great impact on understanding chemistry concepts; inservice junior and senior

students' teachers who studied methodology course show the best understanding for chemistry concept; the

results also show that inspite that freshman science college students had a higher score on General Secondary

Exam upon admission, however, freshman inservice teachers show better understanding for chemistry concept in

the test (both groups did not study methodology course).

Key words: Social, motivation, cognitive, pedagogy, content knowledge, chemistry problems.

- Introduction

One of the important objectives of science education is to develop students' abilities to reason logically; such goal

needs an understanding for the multidimensional of the problem that is: social motivational aspect, the pedagogy

theory, the cognitive aspect, the core conceptual content, and the way to be represented.

- Social Motivational Aspect

Vygotsky ( ) show that human conscious have their origins in human social life, deriving from internalized

social relations that have become functions for the individual and form his structure; he postulated that mental

functioning such as thinking reasoning, problem solving occurs first between people in social interaction,

individuals' mental processes have specific organizational properties that reflect those of the social life from

which they derive.

In the same aspect Gee ( ) pointed out that the mental structure of individual is not static, the social

interaction has the greatest role in the dynamic process of learning which is represented in the transferring from

verbal into written culture -solving science problems is part of such written culture.

Gagne ( ) considers that the motivation has a great role in learning process; it exerts its effects by a means of

a set of expectancy established prior to the act of learning; activation of motivation appealing to student' interest

leads to arousing attention.

Strike and Posner ( ) pointed out the importance of motivation and higher achievement in learning process,

which is contracted with their previous point of view about static Piagean constructivist learning model.

- Pedagogic Theoretical Aspect: Conceptual Change

Loughran et al., defines PCK as the knowledge that a teacher uses to provide teaching situations that help

learners make sense of particular science content. © Centre for Promoting Ideas, USA  
www.ijbhtnet.com

Shulman, L, states that pedagogical content knowledge represents the blending of content and pedagogy and

how particular issue or problems are organized, the manner in which such content is communicated to student,

and what is essential about subject and what is peripheral; Radford, ( ) show that PCK has been successful in

improving the science content knowledge.

Stofflett, & Stoddart, ( ) consider conceptual change pedagogy to be important to get red from previous

misunderstanding for learning experience. Yager ( ), stated that teacher education must involve conceptual

change on the part of teachers, thus making the role of the teacher, as a facilitator of problem solving in science

more crucial.

Gabel, Sherwood, & Enochs ( ) examine the problem –solving skills of high school students and concluded

that one way of helping students overcome the algorithmic mode is to make certain that students understand the

chemical concepts qualitatively before they are represented quantitatively.

- Content Aspect -Crucial Conceptual Content in Chemistry:

Staver ( ) found that students harbor misconceptions about the mole that hinder problem solving, science

textbooks fail to link the mole concept with the concept of standard number of particles, students believe that such

concept is just associated with gram-molecule. Lawrenz ( ), show that fewer than teachers responded

correctly about mass relates to air, motion, and collision. Gabel, Samuel, and Hunn ( ) show that students

have difficulties in understanding the properties of conservation of particles and the orderliness of particles.

Gable, indicates that an emphasis on the particulate nature of matter led to an increase in the overall

achievement scores. Bryan, & Abell, ( ) focuses on presentation of gas properties in terms of the

qualitative-quantitative mode. They show that current pedagogic practice involves minimal use of qualitative

relationships of the gas laws. Staver & Lumpe ( ) state that instruction should place greater emphasis on

molecular representations and relate these representations to the macroscopic and symbolic level. Johnstone

( ) states that sub microchemistry involves particulate ideas, micro representational (symbolic), and

mathematical manipulation, instruction must link the three basic representation of chemistry so that students work

with a combination of the macroscopic, molecular, and symbolic representational modes.

- Cognitive Aspect; Transition from Concrete to Formal

Binns, & Watters, show that large percentage of adult individuals still function at the concrete operational

level; this fact explain the reason that many teachers have misunderstanding of some scientific concepts. Lawson

& Wollman, , show that only of adolescent and adult subjects of their sample were able to make affect

transition to formal cognitive functioning; they examined the effect of instructional procedures that incorporate

ideas designed and employed to successfully affect transition from concrete to formal cognitive functioning. The

results of the study show that such approach encourage orientation towards problem solving, this fact is the

outmost importance for educators. Lawson et al., ( ), point that reasoning by analogy, plays a central role in

the formation of theoretical concept, the concrete operational level of thought for students would be activated in

this way; students gained most when they were taught by teachers who were classified as concrete operational as

opposed to formal operational.

#### - The Problem

The multi dimensional aspects that educators should take into consideration to accomplish better results in

teaching science had been reviewed. The study proposes that learning is a dynamic process that is affected by the

change in expectation, in such a way, that in service students' teachers would have different aspect about learning

from younger and little experience freshman students.

#### - Sample and Procedure

Investigating the succeeding in solving chemistry' problems was examined on three groups of university students:

freshman science college students, n ( ), freshman inservice students' teachers n ( ), junior and senior inservice

students' teachers, n ( ), in one of Jordanian universities. All students study general chemistry course in first

year upon their admission. Inservice students' teachers study methodology course in their second year of

admission. Students had been tested at the end of second semester of the academic year. The freshman students in

Science College had been admitted directly to university programs; they usually had higher score in GSC

(General Secondary Certificate) than in service teachers who had a diploma college certificate in teaching before

being admitted to university programs (diploma equal semesters of university courses). International Journal of Business, Humanities and Technology  
Vol. No. ; March

Instruction in methodology course implies discussing misunderstanding of natural science concepts. The

instruction depends on using analogy as a mediator for transition from concrete to formal understanding. The

even distribution of particles in gas phase had been illustrated by analogy, that the same number of particles gets

the same space regardless of their atomic mass; students would see that passengers should get the same

distribution of seats regardless of their body mass.

- The Instrument

The study used Symbolic Application Particulate chemistry test (SAP), that Dorothy Gable, had designed to

define the conceptual level of high school chemistry teachers in U.S.A. This instrument consists of questions

of problem solving for chemical concepts; density, mixture, conservation of matter, kind of reactions, mole,

chemical reaction, solution, equilibrium, and pH, on three levels; Symbolic, Particulate and Application.

- The Result

The data (represented in Table ( )) show that inservice mature students' teachers who get pedagogic content

knowledge about particulate nature of matter show the best understanding for chemistry concept (average

/ = ) than freshman inservice teachers ( / = ) and freshman Science College Students

( / = ). The results also show that the social motivation, cognitive, and pedagogic dimensions have a

great impact on understanding chemistry concepts, inservice mature freshman students' teachers average made

better than freshman College Science students ( > ), in spite of being admitted younger and with a higher

General Secondary Score. The results also show the importance of instruction that focused on introducing a

pedagogy content knowledge to structure students' misunderstanding in core scientific concepts; inservice junior

and senior students' teachers who get methodology course did better than freshman inservice students' teachers

( > )

- Conclusion

The result of the study shows that learning is a dynamic process affected by the social and motivational aspect;

although freshman Science College students had been admitted younger and with a higher average on General

Secondary Exam, however, their average score on problem solving chemistry test was slightly less than freshman

inservice students' teachers ( < ); the results show the importance of the social aspect of the profession in

changing teachers expectancy from science instruction.

The study also shows that restructuring student' conceptions in the core scientific concepts that have

misunderstanding for it, would lead to better capabilities in solving scientific problems. The emphasis of

instruction on making students represents the abstract concept visually through analogy could explain that effect.

## References

Bryan, L., Abell, S., ( ) "Development of professional knowledge in learning to teach elementary science,

Journal of Research in Science Teaching, ( ) -

Gabel, D., Samuel, K., and Hunn, D.,( )" Understanding the particulate nature of matter" Journal of

Chemical Education, , - .

Gable, D., Sherwood, R., &Enochs, L., ( ), "Problem solving skills of high school chemistry students, Journal

of Research in Science Teaching, ( ), - .

Gabel, D., ( ), Use of particulate nature of matter in developing conceptual understanding, Journal of

chemical Education, , -

Gagne, R.,( ) "The Psychology of Teaching Methods", Yearbook of the National Society for the Study of

Education, Part , Chicago University of Chicago Press, p -

Gee, J. ( ). What is Literacy? In C. Mitchell & Weiler (Eds.). New York : Bergin & Cavey Press.

Ginns, I., & Watters, J., ( ) "An analysis of scientific understanding of preservice elementary teacher

education students" Journal of Research in Science Teaching,, ( ) -

Johnson, P., ( , "Progression in children' understanding of a basis particle theory, Longitudinal study";

"International Journal of science Education", , - .

Johnstone, A ( )," The development of chemistry teaching: A changing response to changing demand", "



Journal of chemical education", , - .© Centre for Promoting Ideas, USA  
www.ijbhtnet.com

Lawrenz, F., ( ) "Misconceptions of physical science concepts among elementary school teachers", Journal of  
School Science and Mathematics, , -

Lawson, A., Wollman, W., "Encouraging the transition from concrete to formal cognitive  
functioning-An  
Experiment. Journal of Research in Science Teaching, , s -s

Lawson, A., Abraham, M., & Renner, J., ( ) A theory of instruction: Using the learning cycle to  
teach science  
concepts and thinking skills. NARST Monograph No. .

Loughram, J., Milory, P., Berry, A., Gunstone, R., & Mulhall, P., ( ) "Documenting science  
teachers'  
pedagogical content knowledge through PaP-eRs". Research in Science Education, , - .

Shulman, L., ( ) "Knowledge and teaching: Foundation of the new Reform, Harvard  
Educational Review  
( ).

Staver, J., ( ), "An analysis of students' errors on an examination question that assessed  
their knowledge of  
the relation between atomic/molecular and molar masses of substance". Paper presented at  
NARST, San  
Francisco, Ca.,.

Staver, J., & Lumpe, A. ( ), "A content analysis of the presentation of the mole concept in  
Chemistry  
textbooks" Journal of Research in Science Teaching, ( ), - .

Stofflett, R., Stoddart, T., ( ) The ability to understand and use conceptual change pedagogy  
as a function of  
prior content learning experience. Journal of Research in Science Teaching, ( ) -

Strike, K., & Posner, G., ( ) "A revisionist theory of conceptual change". In R. Duschl & R. Hilton (Eds.),

Philosophy of science, cognitive psychology, and educational theory and practice (pp. - ). Albany,

NY: SUNY Press.

Vygotsky, L., ( ) "The genesis of higher mental function" in J. Wertsch (Ed) The Concept of Activity in Soviet

Psychology, Armonk, N, Y : Sharp, .

Yager, R., ( ) "The constructivist learning model" The Science Teacher, , -

Table ( )

Students' group No of students Average Percentile Std-deviation

Freshman College students / =

Freshman in service stud- teach. / =

Jun-senior in service stud-teach / =